

Buffer solution pH = 13,00 (20 °C) (Potassium chloride/Sodium hydroxide)**Identification**

Taric code: 3822 00 00

Applications

analytical chemistry, to fit pH of the reaction media.

Specifications

pH at 20 °C..... 13,00

uncertainty ± 0,02

Composition per litre is 3,73g Potassium chloride and 1,92g Sodium hydroxide.

Standard buffer solutions are prepared using gravimetric and volumetric procedures.

The batch value is determined by measurement with a combination glass electrode against two-point calibration according to DIN 19268.

This pH buffer solution is traceable to Standard Reference Material from NIST.

| T (°C) | pH |
|---------|-------|
| 0..... | 13,80 |
| 5..... | 13,59 |
| 10..... | 13,37 |
| 15..... | 13,18 |
| 20..... | 13,00 |
| 25..... | 12,83 |
| 30..... | 12,67 |
| 35..... | 12,59 |
| 40..... | 12,41 |
| 45..... | 12,28 |
| 50..... | 12,15 |

Packaging**Packaging Code**

SO11431 Ⓛ SO11430250

1 l Ⓛ SO11431000

Physical data

- Density: ~ 1,00 g/cm³
- Solub. in water: (20 °C): miscible

Toxicological data

- WGK: 0
- Poison class CH (Swiss): F